

(4)

3. Write short notes on :

(i) axial glide

(ii) isogonal symmetry group [2½×2]

4. Define space group. State the principle of notation of a space group. Describe the space groups under monoclinic system.

[1+1+2]

5. A compound with molecular weight 868 is found to crystallize in monoclinic space group (P2₁/c), which has the following parameters :

a = 9.718 Å, b = 17.469 Å, c = 23.109 Å, β = 91.864° and Z = 4.

Find out the density of the crystal in gm/cm³. 2

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Ex/SC/CHEM/PG/CORE/TH/XI-I/2025

M. Sc. CHEMISTRY EXAMINATION, 2025

(3rd Semester)

PAPER : XI-I

Inorganic Chemistry Special
(X-ray Crystallography + IR, Ramam, CD and
ORD Spectroscopy)

Time : Two Hours

Full Marks : 40

(20 marks for each Unit)

Use a separate Answer Scripts for each Unit.

UNIT—I-3112

Answer any **four** question :

1. (a) Explain Stokes and anti-Stokes shift in Raman scattering from classical theory. 3

(b) What will the intensities of Stokes and anti-Stokes shift in (i) Vibrational-Raman scattering (ii) Rotational Raman scattering? Give reason. 2

(2)

2. Nitrous oxide gives the following IR and Raman spectroscopic data.

$\bar{\nu}(\text{cm}^{-1})$	Infra-red	Raman
589	Strong, <i>PQR</i> contour	—
1285	Very strong, <i>PR</i> contour	Very strong, polarized
2224	Very strong, <i>PR</i> contour	strong, depolarized

- (i) Predict the structure of Nitrous oxide from the above data. 1
- (ii) What do you mean by strong or very strong, *PR* contour term from IR spectroscopic data? 2
- (iii) What do you mean by very strong, polarized and strong, depolarized term from Raman spectroscopic data? 2
3. (a) In quantum mechanics energy of zero vibrational level is not zero. Comment. 1
- (b) Give idea about the frequency of hot band, fundamental band, 1st overtone and 2nd overtone in IR spectroscopy. Discuss the effect of temperature on intensities of those bands. 2+1=3
- (c) Comment on $\nu(\text{O}_2)$ trend of the following species : O_2^+ , O_2 , O_2^- and O_2^{2-} 1

(3)

4. (a) Comment on $\nu(\text{CO})$ values of $\text{Mn}_2(\text{CO})_8\text{X}_2$ compounds where $\text{X}^- = \text{Cl}^-, \text{Br}^-, \text{I}^-$. 2½
- (b) How will you prove the occurrence of linkage isomerism in $[\text{Pd}(\text{dmsO})_4]^{2+}$ (dmsO= dimethylsulfoxide) with the help of IR spectroscopy? 2½
5. (a) What will be the nature of cotton effect in case of diastereomers and enantiomers? 1
- (b) CD (circular dichroism) is used for monitoring the changes in conformation of a nucleic acid as a function of solvent. Explain with a suitable example. 2
- (c) How can we distinguish different conformers of nucleic acid using CD spectroscopy? 2

UNIT—I – 3111

CO1

1. What is a crystal system? Describe the different crystal systems in terms of unit cell parameters and indicate the number of variables for each system. [1+2+1½]
2. State the meaning and draw stereographic projection of any **three** of the following :
(i) *mmm* (ii) 32 (iii) $\bar{4}$ (iv) $\bar{6}2\text{m}$ [1½×3]